Growth of ternary $\text{Si}_{1-x-y}\text{Ge}_x\text{C}_y$ thin films from a single-source precursor, $\text{Ge(SiMe}_3)_4$

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$\text{Ge(SiMe}_3)_4$ was used as a single-source precursor to deposit thin films of alloys of germanium, silicon, and carbon, $\text{Si}_{1-x-y}\text{Ge}_x\text{C}_y$, by low-pressure chemical vapor deposition on silicon substrates at temperatures 873–973 K. X-ray diffraction studies indicated that the films grown above 898 K were cubic phase ($a = 0.441–0.442$ nm). Infrared spectra of the films showed a major absorption near 783 cm$^{-1}$. X-ray photoelectron spectra of a typical thin film showed binding energies of $\text{Ge}_{3d}$, $\text{Si}_{2p}$, and $\text{C}_{1s}$ electrons at 30.0, 100.6, and 283.2 eV, respectively. As determined by wavelength dispersive spectroscopy, $x$ was 0.07–0.15 and $y$ was 0.43–0.50, indicating that the films contained 7–15% Ge, 38–43% Si, and 43–50% C. At 973 K, the $\text{C}/(\text{Si} + \text{Ge})$ ratio was 1. Based on these data, the films deposited above 898 K have a structure of $\beta$–SiC with Ge atoms replacing some Si atoms in the lattice.

In a recent report, Soref showed that in theory, the indirect band gaps of ternary semiconductor materials $\text{Si}_{1-x-y}\text{Ge}_x\text{C}_y$ can span a wide range of 0.62–5.5 eV by adjusting the variables $x$ and $y$. This makes the alloys potentially useful for many applications. Thin films of $\text{Si}_{1-x-y}\text{Ge}_x\text{C}_y$ can be grown by molecular beam epitaxy (MBE) and chemical vapor deposition (CVD). Based on our previous experience of depositing $\beta$–SiC thin films from single-source precursors, employing organopolysilanes of small molar mass as the sources for both silicon and carbon atoms, we think that the alloy films can be grown from molecules containing Ge, Si, and C atoms. In this communication, we report our preliminary investigations on the use of an organometallic compound, $\text{Ge(SiMe}_3)_4$, as a single-source precursor to grow $\text{Si}_{1-x-y}\text{Ge}_x\text{C}_y$ thin films by CVD.

$\text{Ge(SiMe}_3)_4$ was synthesized according to a literature route. After Si substrates in a low-pressure hot-wall CVD reactor were etched by hydrogen plasma to increase nucleation sites, the solid precursor was sublimed at 313 K to start deposition. Deposition of thin films on silicon substrates was performed at 873–973 K and 1 Pa (no carrier gas) and 70 Pa (flow of Hz at 20 sccm). Soft, grey to brown, and mirror-like thin films with good adhesion to the substrates (Scotch Tape test) were obtained. The surface morphology was smooth with very fine grains as shown by the scanning electron micrograph (SEM) in Fig. 1(a). Growth rates, estimated from the cross-sectional SEM micrographs [Fig. 1(b)], were 4–13 nm min$^{-1}$.

The x-ray diffraction (XRD) pattern of a typical thin film grown at 978 K is shown in Fig. 2. It shows major Cu Ka peaks at angles $2\theta$ equal to 35.12°, 40.82°, 59.51°, and 70.79° for the film. These peaks are characteristic of (111), (200), (220), and (311) reflections of a cubic structure with a lattice parameter $a = 0.4417$ nm. The data are close to the values of $\beta$–SiC, 35.59°, 41.38°, 59.98°, and 71.77° for the reflections, and 0.4358 nm for the lattice parameter. Films deposited at 878 K were amorphous as indicated by their XRD patterns. Infrared spectra of the films showed a strong absorption near 783 cm$^{-1}$, which is near but shifted a little from the characteristic Si–C stretching frequency ($\nu_{\text{SiC}}$) of $\beta$–SiC at 800 cm$^{-1}$. No other absorption bands were clearly observed.

Analyses by the following methods confirm that Ge, Si, and C atoms were in the films. High resolution x-ray photoelectron spectra (XPS) of a typical sample deposited at 898 K showed binding energies of the following electrons, $\text{Ge}_{3d}$ (30.0 eV), $\text{Si}_{2p}$ (100.6 eV), $\text{C}_{1s}$ (283.2 eV), and $\text{O}_{1s}$ (532.0 eV), after some of the contaminated surface layers were removed by Ar$^+$ sputtering. The binding energies of the $\text{Si}_{2p}$ and $\text{C}_{1s}$ peaks are close to the values known for $\beta$–SiC, while the $\text{O}_{1s}$ peak is assignable to Si-bonded O atoms. The $\text{Ge}_{3d}$ signal indicates that the Ge atoms were neither condensed into their elemental form, which is at 29 eV, nor connected to hydrocarbon fragments alone, which is at 31 eV. Thus, it is more likely that the Ge atoms were mixed homogeneously with the Si and C atoms. Auger depth profiling showed uniform distributions of Si, Ge, and C atoms within the films. Oxygen concentration was low except at the surface. The films were analyzed by wavelength dispersive spectroscopy (WDS) to determine their bulk compositions. The results are plotted against the temperature of deposition and shown...
in Figs. 3 and 4. For the films deposited without carrier gas (Fig. 3), C concentrations were relatively constant at 46–50%, while Si concentrations increased slightly from 38% to 43% and Ge concentrations decreased from 13% to 7% with increasing temperature of deposition from 873 to 973 K. For the films grown under hydrogen (Fig. 4), Si concentrations were relatively constant at 42–43%, while C concentrations increased from 43% to 50% and Ge concentrations decreased from 15% to 7% with increasing temperature of deposition. In Fig. 5, C/(Ge + Si) ratios are plotted against the temperature of deposition between 873 and 973 K. For the films grown without carrier gas, the ratios span a narrower range of 0.85–1.00 while the ratios for those deposited under hydrogen increase with increasing temperature of deposition from 0.76 to 1.00. However, the difference is not significant enough to draw a conclusion on the effect of hydrogen at this stage. In general, the films grown at high temperatures and without carrier gas have the C/(Ge + Si) ratios close to 1.

Based on the data discussed above, we suggest that the films have a thermodynamically favored β-SiC lattice with some Si atoms replaced by larger Ge atoms. Therefore, the films showed a larger and r_{SiC} smaller than those of β–SiC. It is interesting to note that the temperatures employed to deposit thin films in this study, 873–973 K, are much lower than the conditions required to deposit crystalline β–SiC thin films from organopolysilanes, above 1273 K.\textsuperscript{4–6} This is probably due to a difference in bond strengths between the Ge–Si bonds of Ge(SiMe\textsubscript{3})\textsubscript{4} and the Si–Si bonds of other organopolysilanes. In general, the strengths of M–X bonds (M: group IVA elements; X: other elements) decrease from Si to Pb.\textsuperscript{12} As a result, precursors containing Ge start to decompose at temperatures lower than the decomposition temperatures of analogous Si-containing precursors. For example,
with increasing temperature of deposition, while the precursor has a Ge/Si ratio of 0.25. Formation of tetramethylgermane in the process can be employed as a reason to account for the partial lowering of Ge concentration at high temperatures of deposition. It is not surprising to observe that the films were rich in C since the precursor contains many methyl groups.

In conclusion, we have shown that organometallic compound Ge(SiMe₃)₄ can be employed as a single-source precursor to grow Si₁₋ₓ₋ₓGeₓCy thin films (x = 0.07–0.15, y = 0.43–0.50) by CVD. As suggested by this study, it is possible to use organometallic compounds containing different numbers of Si, Ge, and C atoms as single-source precursors to adjust the composition of Si₁₋ₓ₋ₓGeₓCy thin films grown by CVD. Studies related to this report are in progress.

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