Self-Assembled Ni Nanodot on SiO₂ Film—A Novel Reactive Ion Etching Mask for Si Nanopillar Formation on Si Substrate

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ABSTRACT

By rapid thermal annealing the Ni film evaporated on thin SiO₂ layer covered Si substrate, we have successfully demonstrated the self-aggregation of two-dimensional randomized Ni nano-dots on Si wafer. The thin oxide layer prevents the formation of NiSi₂ compounds and facilitates the self-assembly of Ni nanodots from retaining the thermal power on SiO₂ layer. This greatly shrinks the annealing time required for metallic nanodot formation from >10 min to <30 sec. With the advantage of the self-assemble Ni/SiO₂ nano-dots based nano-mask, a large-area Si nano-pillar array with rod size of <50 nm can be formatted on Si substrate through the induced coupled plasma reactive ion etching (ICP-RIE) procedure. After removing Ni dots and the SiO₂ film on the Si substrate, both the visible and near infrared photoluminescence from the Si nano-pillar sample were observed and analyzed.

Keywords: Self-assemble, Ni nano-dot, Si nano-pillar, reactive ion etching, mask, SiO₂, Si.

1. INTRODUCTION

A Si nano-pillar array has been shown to have the potential for many applications such as being photonic crystals¹, data storage², field electron emitter³, nanoscale transistors⁴,⁵ and light emitting devices⁶. In particular, the light emitting efficiency could be enhanced with the aid of the quantum confined Si nano-pillars. Traditional approaches for fabricating sub-micron or nano- Si structures, such as the Si nano-pillars, mainly rely on the electron-beam (E-beam) lithography¹,⁷-¹⁰, which provides accurate size and shape control. The production of <10-nm nano-pillar array under the assistance of E-beam lithography has been reported by Toshihiko and Papadimitriou. However, it is impractical to employ the E-beam lithography for large-scale fabrication due to its high cost and low writing speed. Nowadays, noble metal (Au or Ag) based nano-dot arrays have been extensively used as nano-masks to replace the E-beam lithography. Recently, Ni has been considered as an alternative to the noble metals. The Ni nano-dot mask is not only easy to control its size but also cheap in its formation process as compared to that of the e-beam lithography. Previously, Yoo et al. made use of Ni nano-dot mask to form Si nano-pillars with diameter and height of about 41nm and 472nm, respectively. Nevertheless, the disadvantage of their results is that the density is too sparse to enhance the luminescence, and the self-aggregation of the Ni nano-dot from a Ni film coated on a Si substrate usually takes up to 10 min even by a rapid thermal annealing (RTA) process at 700°C in the N₂ ambient¹¹. This is mainly due to the larger thermal conductivity of the Si substrate that reduces the thermal energy accumulating on the interface of Ni and Si, and due to the relative good adhesion between the Ni and Si interface that slows down the aggregating speed. Besides, such a long-term annealing inevitably leads to an unintentional doping or diffusion of the coated metal into the Si substrate during the self-assembly procedure. To avoid the aforementioned problems, we propose for the first time the quick formation of Ni nano-dots on Si substrate covered with a thin buffered SiO₂ layer, Si₃N₄ layer¹² or TiN¹₃,¹⁴. The thin oxide or nitride layer could prevent the formation of NiSi₂ compounds and keep the thermal power on the Si surface. The reactive ion etching and electrochemical etching were employed as two etching procedures to obtain Si nano-pillar array with the aid of self-assembled Ni nano-dot mask patterns. With the novel approach, the formation of the Si nano-pillar array can be mostly realized by the reactive ion etching of oxide-covered Si substrate capped with the mask of self-assembled Ni, Al¹⁶ or Au¹⁷ nano-dot. Afterwards, the luminescent characteristics of the Si nano-pillars or nano-rods generated at the reactive ion etched Si surface are investigated.
2. EXPERIMENTAL

The schematic diagram to perform the experimental steps of the Si nano-pillar formation is shown in Fig. 1, which includes wafer cleaning, SiO₂ deposition, Ni evaporation, rapid thermal annealing for Ni nanodot assembly, ICP dry etching for Si nano-pillar, and wet etching for metal/oxide removal. First of all, the RCA cleaning process is also employed to completely clean the dusty particles and native oxide of a p-type [100] Si wafer, which flushes the Si wafer with DI (de-ion) water, immerse the Si wafer in the H₂SO₄ based solution to decompose and oxidize the organic compounds, dip it into the HF solution to get rid of chemical oxide, immerse it in NH₄OH solution to get rid of micro particles, dip it in HCl solution to get rid of alkaline-earth metal, and finally dip it in HF to get rid of chemical oxide.

In previous experiment, it is hard to achieve the aggregation of evaporated Ni film on pure Si wafer. This is because the adhesion between Ni and Si is too good to be make Ni self-assembled. Such a problem can be solved by depositing a thin layer of SiO₂ between Ni and Si. In our case, a buffered layer of 200Å-thick SiO₂ is deposited by plasma enhanced chemical vapor deposition (PECVD) under standard recipe. Afterwards, a 50nm-thick Ni film is evaporated on the SiO₂/Si substrate using an E-beam evaporating system with Ni deposition rate of 0.1 Å/s at an applied current of 70 mA. Subsequently, the rapid thermal annealing (RTA) process at 850°C for 22 seconds under the N₂ flowing gas of 5 sccm is performed to format randomized Ni nanodot pattern on SiO₂/Si substrate. By using the Ni nano-dot pattern as an etching mask, The Si substrate is dry-etched in a planar type ICP-RIE system (SAMCO ICP-RIE 101iPH) at RF frequency of 13.56 MHz and the ICP/Bias power conditions of 100/50 or 200/100 watts. The ICP has a reactive chamber connected to a load-lock chamber, wherein the etching gas mixture of CF₄ and Ar gases were introduced into the reactive chamber through individual electronic mass flow controllers (MFCs). The flowing rate of each gas is adjusted within an accuracy of about a standard cubic centimeter min (sccm). An automatic pressure controller is placed near the exhaust end of the chamber to control the chamber pressure. The gas mixture condition of CF₄/Ar = 40/40 sccm with the ICP/Bias powers set at 100W/50W or 200W/100W and chamber pressure of 0.66 Pa during an etching duration of 5~7 min. If the etching parameters such as source power, bias power, and pressure and gas composition are investigated and optimized, Si nano-pillar arrays with tunable diameters of smaller than 100 nm can be formed. For analyzing the morphology of Ni nano-dot, the dimension and density of the Si nano-pillar array were estimated by the scanning electron microscope (SEM, Hitachi FE-SEM S-5000) and the standard atomic force microscopy (AFM).

3. RESULTS AND DISCUSSION

3.1 Effect of SiO₂ Buffered Layer

Figure 2 is SEM image of the Ni nano pattern under different thickness of SiO₂ layer when the Ni thickness, annealing temperature and time are 50 Å, 850°C and 2 min, respectively. After RTA, Fig. 2(a) confirms that there is not any aggregation of Ni on pure Si wafer at all. On the contrary, we observe the apparent self-assembled Ni dot on SiO₂/Si substrate. The image in Fig. 2(b) shows that the Ni layer begins to form a long strip shape on the thin SiO₂ layer after annealing. By increasing the thickness of SiO₂ layer from 100 Å to 200 Å, Fig. 2(c) illustrates a coherent aggregation of Ni nanodots with almost half-sphere morphology. If the thickness of the SiO₂ enlarges to 1000 Å, the size of Ni nanodots becomes larger than those obtained with thinner SiO₂ film covered on the Si substrate, whereas the Ni nanodot
density greatly reduces. It is concluded from the above-mentioned comparison that the congregation of Ni can be more complete as the thickness of SiO\(_2\) increases, which is mainly attributed to the enhanced heat accumulation effect happened in the thicker SiO\(_2\) layer. Apparently the conductivity of Si is better than SiO\(_2\) due to their distinguishable thermal conductivities of 148 and 1.31 W/m-K, respectively\(^{15}\). Therefore, the SiO\(_2\) can lay off the thermal diffusion from Ni layer to the bowl and also can accumulate thermal energy dissipated from Ni layer to help Ni self-assembled.

![Fig. 2 Aggregation of Ni nanodots on SiO\(_2\) layer with different thicknesses ranging from 100Å to 1000Å depositing on Si substrate](fig)

3.2 Effect of Annealing Temperature and Period
The thin Ni film evaporated on the SiO\(_2\) film can easily to be conglomerated, resulting in the formation of Ni nano-dot array on SiO\(_2\) substrate. Whether we change the annealing temperature or annealing time, both the thermal energy of annealing can be adjusted. From our experimental results, 850\(^{\circ}\)C is the proper temperature for RTA annealing. As if the annealing temperature is too high, the Ni and Si atoms could obtain sufficient power to compose the SiNi\(_2\) compounds\(^{15}\). However, if the annealing temperature is too low, it spends more annealing time to congregate Ni film into randomized two-dimensional nanodot array. In the following experiments, we choose to fix the annealing temperature and vary the annealing time during RTA process.

![30(s) and 120(s)](fig)
Fig. 3 SEM image of Ni dots after different annealing time at 850°C. Apparently the density of dots for longer annealing time is denser.

As shown in Fig. 3, we compare the variation in size and density of Ni dots assembling on 1000Å SiO₂ layer under two different annealing time. The average size and density of Ni nanodots annealing at 850°C for 30 sec are 37 nm and 1×10¹² cm⁻², respectively. Nonetheless, the average size and density of Ni nanodots annealing at 120 sec significantly increase to 40 nm and 7×10¹¹ cm⁻². This again confirms the trend of enlarging size and decreasing density under long annealing time. The Ni atoms can get more thermal energy to complete their precipitation process under longer annealing time. However, after rapid thermal annealing at 850°C between 20 and 26 sec, the Ni dots were formed on the 200Å SiO₂ film shown in Fig. 4.

![Ni dots SEM images](image)

Fig. 4 The size and density of Ni dots as a function of the annealing time when the thickness of SiO₂ is 200Å. The 22 sec annealing time at 850°C is the optimum time for the densest density and the smallest diameter.

The shape of Ni nanodots becomes hemispheric-like with an average diameter varying from 33 nm to 75 nm, respectively. A largest density of 7.2×10¹² cm⁻² for the Ni nanodot with smallest size of 33 nm can be obtained after annealing at 850°C for 22 sec. Figure 4 interprets the variation SEM image of Ni nanodot self-assembled at gradually lengthened annealing durations. The Ni film initially breaks into large strips and slowly approaches a hemispheric shape with increasing RTA time, eventually these small Ni nanodots would congregate to larger nanodots if the annealing time prolongs to >30 sec. At annealing time between 22 and 26 sec, there is also a similar tendency on the size and density of Ni nanodots as compared to those shown in Fig. 3.
In more detail, the size and distribution of the Ni nano-dots on the SiO₂ film are mainly determined by thin oxide film thickness, RTA temperature, RTA time and Ni thickness. From the diagram of curves of Fig. 5, the optimum time is 22 seconds. The optimum annealing condition for Ni nanodot formation with highest density and smallest size are therefore set as 850°C for 22 sec in our RTA annealing process. Such a sample has also been chosen as the Ni nanomask for etching Si-nano-pillars. In addition, we also observe that the diameter of Ni nanodots becomes smaller with a decrease in the thickness of the deposited Ni layer.

3.3 Effect of Etching Power
To perform the Si nanopillar formation, we took the Ni nanodots as a nano-scale mask that formed at the annealing 850°C for 22 sec. The average diameter and density are 33nm and 7.2×10¹² cm⁻². The effect of processing pressure on the etching characteristics of Si was investigated. The processing pressure was controlled between 6.66 and 0.66 Pa and the other processing parameters are unchanged. Since the chamber with the high processing pressure contains the more reactive gases, the etching rate will be faster than that with lower processing pressure. However, it is difficult to form the Si nanopillars under the high processing pressure. The reason is that the Ni dots were also etched completely. So we always controlled the chamber pressure under 0.66 Pa during the etching.

Figure 6(a) shows that the by-product appears and the shape of Si nano-pillar resembles a pyramid when the ratio is 3. Then the density is the smallest and the diameter is the largest. The cause is isotropic etching erodes the smaller pillar, and the reason for a pyramid shape is that the vertical etching-rate is not bigger than the side etching-rate so the diameter of the pillar becomes bigger after longer etching time. Finally, the etching rate is dominated by the power. So we are sure the ratio of 3 is too big to get a straight pillar. Figure 6(b) shows the success of the formation of Si nano-pillar with a perpendicular sidewall at the ratio of 2. The density and diameter of Si nano-pillars are 2.8×10¹² cm⁻² and 29 nm. Figure 6(c) shows the straight but short Si nano-pillars at the ratio of 1. This is because the vertical etching is too big to balance the etching selection between Ni and Si, we can see the Ni mask disappeared during the etching.
In another word, the appropriate ratio of bias and ICP power is the key point to get sharp sidewall and large height of Si pillar. After analyzing the SEM data in Fig. 7, we can know the ratio of bias power over ICP power had great influence on the character of Si nano-pillars such as the size and the density. In summary, we know that the ratio of 2 is the most suitable for the formation of Si nano-pillar. From Fig. 8 we can know if the magnitude is bigger, the Si nano-pillars will be longer and the size of them will be wider. The average size of bias/ICP power at 100(W)/200(W) is 29 nm and the average one at 50(W)/100(W) is 44nm. The average height of 100(W)/200(W) is 150nm and the one of 50(W)/100(W) is 250nm. And from the cross-section image in Fig. 8, high aspect-ratio of Si nano-pillars was formed at the bias/ICP power of 2.

As for etchant gas, we chose the O₂, CF₄ and Ar at the beginning of experiment. CF₄ is the main gas for etching SiO₂ and Si. Ar is also main gas able to increase the vertical-etching potential. In our opinion, O₂ is the mirror gas which should speed up the etching rate. Nevertheless, the aid of O₂ isn’t apparent and the oxygen plasma causes a lot of defects such weak oxygen bond and the neutral oxygen vacancy in the Si nano-pillars analyzed from the photoluminescence. On account of the above reason, we have not used O₂ as etchant gas.

**3.4 Results of Photoluminescence**
The height of the Si nano-pillars is between 30 and 100 nm shown in Fig. 9(b) and the density of the Si nana-rods is about $5.4 \times 10^{11}$ cm$^{-2}$ shown in Fig. 9(a). Therefore, some Ni dots were eliminated from the surface of the SiO₂ film. The crater on the Si nano-pillars also provides another proof of the Ni-dot elimination. After removing Ni dots and the SiO₂ film on the Si substrate, the visible photoluminescence from the Si nana-rod sample were observed.
Previously, the luminescent centers in the SiO$_2$:Si$^+$ corresponding to the visible PL were comprehensively investigated, which include the weak oxygen bond (WOB), the neutral oxygen vacancy (NOV) defect, the E’$\delta$ defect and the non-bridge oxygen hole center (NBOHC) defects at emitting wavelengths of around, 415 nm, 455 nm, 520 nm, and 630 nm, respectively\textsuperscript{19-22}. The Si nano-pillar sample presents a strong and broad PL spectrum between 400 nm and 600 nm. After decomposing with multi-Gaussian function, two principle luminescent centers at 418 and 451 nm with linewidths of 39 and 84 nm, respectively, are demonstrated shown in Fig. 9(c). The strongest PL peaks at 415-455 nm with linewidths of 35-50 nm are very similar to those obtained by Nishikawa \textit{et al.} The luminescence at 455 nm reported by Bae \textit{et al}. has been attributed to the transition between the ground state (singlet) and the elevated state (triplet) of the NOV defects. During the etching process, oxygen atoms generated from the ionized O$_2$ impact the surface of the Si substrate. Therefore, two oxygen atoms generate the WOB defect. The NOV defect is attributed to the mobile oxygen atom and oxygen vacancy. If we changed the pumping power density, we can observe the changeless peak and the defect peak saturating shown in Fig. 10. This was because the density of the pumped energy level saturated under the high pumping power. This is also a proof that the two Gaussian peaks are from the defects not from the Si quantum confine. According to the information the quantum confine occurred at the size below 10 nano meters. Obviously, we must control the Si nano-pillar size below 10 nm to get quantum effect in the future.

![Fig. 9 (a) SEM image of Si nano-pillars after removing the Ni and SiO$_2$ layers at the etching-power ratio of 2 for 3 min. (b) AFM data including 3-D 45°view angle image and the height profile. (c) Photoluminescence of Si nano-pillars in (a) and decomposed by two Gaussian fits at 418 and 451 nm.](image)

![Fig. 10 (a) Photoluminescence of Si nano-pillars for different pumping power densities. (b) Peak intensity for different pumping power densities.](image)

### 3.5 Comparison with another process
The thin film deposition and RTA process is not only an easy and simple technique to control the formation of nano-dot but also an effective process compared to the conventional nano-pattern formation process like electron beam lithography technique. If the annealing temperature is too high, Ni and Si atoms would compose into SiNi2 compounds. However, longer annealing time is required to congregate Ni film to nanodots if the annealing temperature is too low. 850°C is the most suitable temperature for RTA annealing, and the Ni dot can be formatted just after 20 seconds. In comparison with previous work, the density of Ni dots of our experimental results is 7.2×10^{12} cm^{-2} that is much higher than those ever reported (5×10^{11} cm^{-2})\textsuperscript{11}. The average size of 33 nm for the obtained Ni nanodots is also far smaller than the previous record of 60 nm. Moreover, the uniformity of the Ni nanodots obtained from our condition is clearly better than the previous works. Most important, we greatly reduce the annealing time from 10 minutes to 22 seconds with the aid of a buffered SiO2 layer.

In the ICP-RIE process, the use of O2 source gas is abandoned in our recipe since which causes the oxygen related defects of such O2-, oxygen vacancy, weak-oxygen bonds, and non-bridged oxygen hole centers in Si matrix. Although the aspect ratio of our Si nano-pillars is only 5 as compared to the results reported by Homma et al.\textsuperscript{17} (up to 7) and Lee et al. (up to 12)\textsuperscript{11}. In our case, the high-aspect-ratio selective etching procedure can be improved by increasing the etching time and by optimizing the ratio of bias/RF power and Ar fluence. The average diameters of Si nano-pillars obtained in our results of 29 nm and 54 nm from different recipes is larger than the record of <10 nm by Japan and Korean groups. It tells us our Ni dots don’t disappear during the etching and the etching selective is high to get long and small diameter Si nano-pillar. However, the density of our result (a largest density of this process can be nearly 7×10^{12} cm^{-2})\textsuperscript{,} by using the dense Ni nanodots based nano-mask is still larger than the records to date.

4. CONCLUSION

A novel reactive ion etching mask for Si nano-pillar formation on Si Substrate with self-assembled Ni nano-dots on SiO2 film was successfully investigated. The improvement of this experiment is shortening the annealing time to 20 seconds and increasing the density of nano-pillars by the dense Ni nano-dots. From the above discussion, we can get the control of the density and size of Ni nano-mask by annealing temperature, annealing time, SiO2 thickness and Ni thickness. If the SiO2 film is thicker, the larger size and less density of Ni dots will be achieved. However, a thick SiO2 layer would decrease the effective etching depth of Si nano-pillars. Therefore, 200 Å-SiO2 is selected to be a buffer layer for Ni assembly. High annealing temperature and long annealing time will cause the large size and less density of Ni dots. Nevertheless, from the beginning of Ni congregation, the maximum density and size of Ni dots are 7.2×10^{12} cm^{-2} and 33 nm, respectively. The optimum annealing temperature and time are 850°C and 22 seconds for Ni congregation. The optimum ICP-RIE recipes for the densest and highest aspect-ratio Si nano-pillar are 0.66 pa and 2 of the low chamber pressure and the ratio of RF power over bias power. The largest density and the average size of Si nano-pillar are 7×10^{12} cm^{-2} and 50nm, respectively. The average height of Si nano-pillar is about 100 nm for the etching time of 3 mins. For the etching time of 5 mins, the average diameter of Si nano-pillars is narrow down to 29 nm with the density of 2.84×10^{12} cm^{-2} and the average height of 130 nm. Moreover, an obvious PL between 350 nm and 600 nm can be observed. After decomposing with the multi-Gaussian function, two principle luminescent centers at 418 and 451 nm with spectra linewidths of 39 and 84 nm, respectively, are demonstrated. Since the PL intensity is saturated under the high pumping power, Gaussian peaks are attributed to defects, such as WOB and NOV defects, not the Si quantum confinement from Si nano-pillars. The future experiment tends to control the Si nano-pillar size below 10nm to get quantum effect.

REFERENCES